

Evaluating the Effectiveness of Solvent Free Removal of Waxes



Consumables

1.0: Purpose

The mounting of specimens for specimen preparation is a common practice in virtually all metallographic, optical, and electron microscopy applications. In most cases the specimens require some type of polishing technique which necessitates the use of a temporary adhesive to fix the specimen during preparation. The use of low melting point wax is the most common technique, involving the heating of the specimen to a temperature at or near the melting point of the wax being used. The specimen is then affixed to a block or mount of some type, followed by a series of grinding and polishing steps to achieve the desired section and surface finish. Typically the specimens are reheated and then cleaned in a solvent such as acetone or methanol. However, with certain types of specimens such as polymers, plastics, and some types of circuit boards, solvent cleaning cannot be used. For these types of specimens a cleaning solution is required which will not attack and dissolve the specimen, or an alternative method for mounting the samples is required. In this experiment, a new wax removal technique will be tested. A compound which is soluble in water and commonly used for semiconductor wafer cleaning will be tested to evaluate the effectiveness of removing wax from a specimen. Different materials will be evaluated to ensure that the wax removal (or lack thereof) is uniform with most material systems. Comparing the new water cleaning method with traditional solvent cleaning will also be examined.

2.0: Experiments and Procedures

Four different specimen types were tested in this experiment. Samples of silicon (Si), quartz, brass and printed circuit board were all heated to an approximate temperature of 130°C and then a small amount (~ 2 mm²) of wax was applied to the surface and allowed to cool. Once the specimen had cooled to room temperature, they were placed into a bath of the water soluble cleaner (~ 20 ml) and ultrasonically cleaned for 5 minutes. The cleaner was set temperatures varying from 65°C up to 100°C prior to cleaning of the specimens. Three different ratios of cleaner were tested to determine any possible anomalies in concentration. Once the specimens were cleaned, they were each rinsed by dipping into a beaker of water 3 times and then dried on filter paper.

Four different wax types were tested in this experiment. Below is a table illustrating the properties and types of waxes used for testing.

Designation	Wax Type	Melting Point (°C)	Strength	Hardness	Solubility
1	MWH 135	135	High	Very Hard	Acetone
2	MWH 080	80	High	Hard	Ethanol, acetone
3	MWM 070	70	Moderate	Moderate	Warm soapy water
4	MWS 052	52	Moderate	Soft	Perchloroethylene

The wax samples were applied to each specimen type, with a total of 16 samples cleaned. The same process was done using acetone as the primary cleaner, followed by a water rinse as described above.

3.0: Results

Following the cleaning tests, a large data set was collected which showed the results of all of the testing done. Each aspect of the experiments will be discussed in detail below. The charts can be evaluated at the end of this report.

Table 1

Table 1 is the data collected for the 3.5 to 1 mixing ratio (water to powder) of the WaxStrip solution. This is the smallest mixing ratio tested, making it the most concentrated solution used in the cleaning process. The solution was heated to a temperature of 100°C with a sonication bath temperature of 30°C. Each of the samples were sonicated for 5 minutes and then the surface was examined optically to help determine the amount of wax removal obtained by the cleaning. From the chart it can be seen that for samples contaminated with wax #1 the removal was virtually non-existent. However, for the other three waxes, #2, 3, and 4, cleaning was very efficient. The variation in cleaning compared to the various sample types is minimal, with all four different samples cleaned equally, depending upon the mounting wax they were contaminated with.

Table 2

Table 2 is the data collected for the 5.3 to 1 mixing ratio of the WaxStrip solution. This is an in between mixing ratio and is less concentrated than the one used in Table 1. The solution again was heated to a temperature of 100°C with a sonication bath temperature of 30°C. Again, as with the results in Table 1, mounting wax #1 was not removed efficiently as compared with the other three waxes. However, there was some removal of wax #1 in the brass and silicon samples observed. The efficiency of the cleaner to remove the other three types of wax seems to be quite high.

Table 3

Table 3 shows the data collected for a 10 to 1 mixing ratio of the WaxStrip solution. The solution was at a temperature of 100°C with a sonication bath temperature of 30°C. The results were virtually identical to those obtained with the 5.3 to 1 mixing ratio, with slight improvement in the removal of wax #1 on the brass and silicon samples. All other waxes were removed relatively easily.

Table 4

Table 4 shows data collected for a cleaning solution at a 10 to 1 mixing ratio, which was observed to be the optimum mixing ratio from the previous experiments. The cleaning solution temperature was reduced to 80°C and the sonication bath temperature was kept constant at 30°C. No significant change in cleaning characteristics were observed with those of Table 3.

Table 5

Table 5 shows the same experiment as Table 4 with a cleaning solution temperature of 70°C and a sonicating bath temperature of 70°C as well. In this case, all four wax types and sample types were cleaned with equal efficiency. The sonicating bath temperature seems to have had a positive effect on the outcome of the cleaning solutions.

Solvent Cleaning

Some experiments were run using acetone cleaning as a pre cleaning step prior to the WaxStrip cleaning. These were shown to be effective in removing wax residue, although the benefit of solvent free cleaning is lost using this method. A slight amount of residue was seen on the silicon samples which was not present from solvent free cleaning, but is unclear as to whether or not this is due to the solvent step or not. Further experimentation may be needed to evaluate this further.

4.0: Conclusions

Based on the experiments and data collected, the use of a solvent free cleaning solution can be quite useful in removal of most mounting waxes. Without the need for solvent cleaning, specimens which are normally affected by solvents can be cleaned as well. The most critical parameters which affect the outcome of cleaned specimens is the mixing ratio of water to WaxStrip powder; cleaning solution temperature and sonicating bath temperature. The optimum conditions are to keep the mixing ratio at 10:1, keep the sonicating bath and the cleaning solution temperature both the same. All of these combined produce well cleaned specimens.

